

17.14 FORMULAE:

(1) Mass Variation

$$m = \frac{m_0}{\sqrt{1 - \frac{v^2}{c^2}}}$$

$m \rightarrow$ Relativistic mass
 $m_0 \rightarrow$ Rest mass

(2) Time Dilation

$$t = \frac{t_0}{\sqrt{1 - \frac{v^2}{c^2}}}$$

$t \rightarrow$ Relativistic time
 $t_0 \rightarrow$ Proper time

(3) Length Contraction

$$L = L_0 \sqrt{1 - \frac{v^2}{c^2}}$$

$L \rightarrow$ Relativistic length
 $L_0 \rightarrow$ Proper length

(4) Mass Energy Relation

$$E = mc^2$$

(5) Kinetic energy of particle in non-inertial frame

(i) $K.E = E - E_0$;
 $E_0 = m_0 c^2$

$E \rightarrow$ Total or Relativistic energy
 $E_0 \rightarrow$ Rest mass energy

(ii) $K.E = \frac{p^2}{2m_0}$;

$p \rightarrow$ Momentum of Particle

(6) Stefan's law

$$E = \sigma T^4$$

$\sigma \rightarrow$ Stefan's constant
 $\sigma = 5.67 \times 10^{-8} \text{ W m}^2 \text{ K}^{-4}$

(7) Raleigh Jean's law

$$E = \frac{\text{Constant}}{\lambda^4}$$

(8) Wein's Displacement law

$$\lambda_{\text{max}} T = \text{constant}$$

$\lambda_{\text{max}} \rightarrow$ wave length of maximum energy radiated

Constant = 0.0029 m.k

(9) Plank's law or Energy of Photon or Quanta or Radiations

(i) $E = h\nu$;

$h \rightarrow$ Plank's constant
 $h = 6.63 \times 10^{-34} \text{ J.S}^{-1}$

(ii) $E = \frac{hc}{\lambda}$

(10) Relation between light speed, wave length and frequency

$$C = \nu\lambda$$

- (11) Work function

$$\phi = h\nu_0 \quad ; \quad \nu_0 \rightarrow \text{threshold frequency}$$

- (12) Kinetic energy of Photo electrons from

- (i) Stopping Potential

$$\frac{1}{2} m_0 v^2 = eV_0 \quad ; \quad V_0 \rightarrow \text{Stopping Potential}$$

- (ii) $\frac{1}{2} m_0 v^2 = E - \phi$

- (13) Cut - off wave length

$$\lambda_c = \frac{hc}{eV_0}$$

- (14) Wave length of minimum energy

$$\lambda_{\min} = \frac{hc}{\phi}$$

- (15) Change or shift in wave length of scattered X - rays

$$\lambda_2 - \lambda_1 = \frac{h}{m_0 c} (1 - \cos \theta) \quad ; \quad \theta \rightarrow \text{scattering angle of X - rays}$$

- (16) De Broglie's wave length or wave length of particle from velocity

$$\lambda = \frac{h}{p} \quad \text{OR} \quad \lambda = \frac{h}{m_0 v}$$

- (17) Wave length of particle in accelerating frame from accelerating potential

$$\lambda = \frac{h}{\sqrt{2m_0 eV}}$$

- (18) Pair production

$$E = 2(m_0 c^2) + (K \cdot E)_{e^-} + (K \cdot E)_{e^+}$$

- (19) Uncertainty Principle

- (i) Momentum - Position Uncertainty

$$\Delta p \Delta x = \hbar \quad ; \quad \hbar = \frac{h}{2\pi} = 1.05 \times 10^{-34} \text{ J.S}^{-1}$$

- (ii) Energy time uncertainty

$$\Delta E \Delta t = \hbar$$

17.15 SOLVED NUMERICALS OF BOOK:

PROBLEMS

17.1: In the inertial frame of Pendulum the time period is measured to be 3 seconds. What will be the period of the Pendulum for an observer moving at a speed of 0.95C with respect to the Pendulum?

DATA:

$$t_0 = 3 \text{ Sec}$$

$$V = 0.95C$$

$$t = ?$$

SOLUTION:

$$t = \frac{t_0}{\sqrt{1 - \frac{V^2}{C^2}}}$$

$$t = \frac{3}{\sqrt{1 - \left(\frac{0.95C}{C}\right)^2}}$$

$$t = \frac{3}{\sqrt{1 - 0.9025}}$$

$$t = \frac{3}{0.312}$$

$$t = 9.6 \text{ sec}$$

17.2: What will be the length of a bar in the stationary frame if it is length along x - direction is 1m and the motion is with a velocity of 0.75C with respect to the observer at rest?

DATA:

$$L_0 = 1 \text{ m}$$

$$V = 0.75C$$

$$L = ?$$

SOLUTION:

$$L = L_0 \sqrt{1 - \frac{V^2}{C^2}}$$

$$L = 1 \sqrt{1 - \left(\frac{0.75C}{C}\right)^2}$$

$$L = 1 \sqrt{1 - 0.5625}$$

$$L = 0.66 \text{ m}$$

17.3 Given $m_0C^2 = 0.511 \text{ MeV}$. Find the total energy and the kinetic energy of an electron moving with the speed $v = 0.85C$

DATA:

$$E_0 = 0.511 \text{ MeV}$$

$$V = 0.85C$$

$$E = ?$$

$$K.E = ?$$

SOLUTION:

$$\text{as } m = \frac{m_0}{\sqrt{1 - \frac{V^2}{C^2}}}$$

$$mC^2 = \frac{m_0C^2}{\sqrt{1 - \frac{V^2}{C^2}}}$$

$$E = \frac{E_0}{\sqrt{1 - \frac{V^2}{C^2}}}$$

$$E = \frac{0.511}{\sqrt{1 - \left(\frac{0.85C}{C}\right)^2}}$$

$$E = \frac{0.511}{\sqrt{1 - 0.7225}}$$

$$E = \frac{0.511}{0.5268}$$

$$E = 0.970 \text{ MeV}$$

$$K.E = E - E_0$$

$$K.E = 0.970 - 0.511$$

$$K.E = 0.459 \text{ MeV}$$

- 17.4 The total energy of a proton of mass 1.67×10^{-27} kg is three times its rest energy. Find
 (a) Protons rest energy. (b) Speed of proton. (c) Kinetic energy of proton in eV.

DATA: $m_0 = 1.67 \times 10^{-27}$ Kg
 $E = 3E_0$

- a) $E_0 = ?$ b) $V = ?$ c) $K \cdot E = ?$

SOLUTION:

a) $E_0 = m_0 C^2$
 $E_0 = (1.67 \times 10^{-27}) (3 \times 10^8)^2$
 $E_0 = 1.503 \times 10^{-10} \text{J}$

OR $E_0 = \frac{1.503 \times 10^{-10}}{1.6 \times 10^{-19}}$

$E_0 = 9.39 \times 10^8 \text{ eV}$

OR $E_0 = 939 \text{ MeV}$

b) $E = 3 E_0$
 $m C^2 = 3 m_0 C^2$
 $m = 3 m_0$

$\frac{m_0}{\sqrt{1 - \frac{V^2}{C^2}}} = 3 m_0$

$\frac{1}{3} = \sqrt{1 - \frac{V^2}{C^2}}$

$\frac{1}{9} = 1 - \frac{V^2}{C^2}$

$\frac{V^2}{C^2} = \frac{8}{9}$

$V^2 = \frac{8}{9} C^2$

$V = \sqrt{\frac{8}{9}} C$

$V = 0.94 C$

c) $K \cdot E = E - E_0$

$K \cdot E = 3E_0 - E_0$

$K \cdot E = 2E_0$

$K \cdot E = 2 \times 939$

$K = 1878 \text{ MeV}$

- 17.5 A Particle of rest mass m_0 has a speed $V = 0.8C$. Find its relativistic momentum, its Kinetic energy and total energy.

DATA:

rest mass = m_0
 $V = 0.8C$

- a) $p = ?$
 b) $E = ?$
 c) $K \cdot E = ?$

SOLUTION:

a) $p = mv$

$p = \frac{m_0 V}{\sqrt{1 - \frac{V^2}{C^2}}}$

$p = \frac{m_0 \times 0.8C}{\sqrt{1 - \left(\frac{0.8C}{C}\right)^2}}$

$p = \frac{0.8 m_0 C}{\sqrt{1 - 0.64}}$

$p = \frac{0.8 m_0 C}{\sqrt{0.36}}$

$p = \frac{0.8 m_0 C}{0.6}$

$p = \frac{4}{3} m_0 c$

b) $E = mc^2$

$E = \frac{m_0 C^2}{\sqrt{1 - \frac{V^2}{C^2}}}$

$E = \frac{m_0 C^2}{\sqrt{1 - \left(\frac{0.8C}{C}\right)^2}}$

$E = \frac{m_0 C^2}{\sqrt{1 - 0.64}}$

$E = \frac{m_0 C^2}{\sqrt{0.36}}$

$E = \frac{m_0 C^2}{0.6}$

$E = \frac{5}{3} m_0 C^2$

c) $K \cdot E = E - E_0$

$K \cdot E = \frac{5}{3} m_0 C^2 - m_0 C^2$

$K \cdot E = \frac{5 m_0 C^2 - 3 m_0 C^2}{3}$

$K \cdot E = \frac{2}{3} m_0 C^2$

17.6: What will be the velocity and momentum of a particle of rest mass m_0 whose kinetic energy is equal to its rest mass energy?

DATA:

rest mass = m_0

$V = ?$

$P = ?$

$K.E = E_0$

SOLUTION:

Given that $K.E = E_0$

$E - E_0 = E_0$

$E = 2 E_0$

$m_0 c^2 = 2 m_0 c^2$

$m = 2 m_0$

$$\frac{m_0}{\sqrt{1 - \frac{V^2}{C^2}}} = 2 m_0$$

$$\frac{1}{2} = \sqrt{1 - \frac{V^2}{C^2}}$$

$$\frac{1}{4} = 1 - \frac{V^2}{C^2}$$

$$\frac{V^2}{C^2} = 1 - \frac{1}{4}$$

$$\frac{V^2}{C^2} = \frac{3}{4}$$

$$V^2 = \frac{3}{4} C^2$$

$$V = \frac{\sqrt{3}}{2} C$$

$$\boxed{V = 0.87C}$$

$P = mv$

$$p = 2 m_0 \times \frac{\sqrt{3}}{2} C$$

$$\boxed{p = \sqrt{3} m_0 C}$$

17.7 The sun radiates energy at a rate of $3.8 \times 10^{26} \text{ W}$. At what rate the mass of the sun diminishes?

DATA:

$P = 3.8 \times 10^{26} \text{ W}$

Time = 1 year = $3.15 \times 10^7 \text{ sec}$

rate of mass diminished per year = ?

SOLUTION:

$$P = \frac{E}{t}$$

$E = P \times t$

$m C^2 = P \times t$

$$m = \frac{P \times t}{C^2}$$

$$m = \frac{3.8 \times 10^{26} \times 3.15 \times 10^7}{(3 \times 10^8)^2}$$

$$\boxed{m = 1.33 \times 10^{17} \text{ Kg/year}}$$

17.8: What will be the work function of a substance for a threshold frequency of $43.9 \times 10^{13} \text{ Hz}$?

DATA:

$\nu_0 = 43.9 \times 10^{13} \text{ Hz}$

$\Phi = ?$

$h = 6.63 \times 10^{-34} \text{ J s}^{-1}$

SOLUTION:

$\Phi = h \nu_0$

$$\Phi = 6.63 \times 10^{-34} \times 43.9 \times 10^{13}$$

$$\Phi = 2.93 \times 10^{-19} \text{ J}$$

$$\Phi = \frac{2.93 \times 10^{-19}}{1.6 \times 10^{-19}}$$

$$\boxed{\Phi = 1.83 \text{ eV}}$$

17.9: What will be the value of $\lambda_{\min} = \frac{hc}{eV_0}$; if $h = 6.63 \times 10^{-34} \text{ Js}^{-1}$, $e = 1.6 \times 10^{-19} \text{ C}$ and $V_0 = 10^4 \text{ V}$, $C = 3 \times 10^8 \text{ ms}^{-1}$

DATA:

$$\begin{aligned} \lambda_{\min} &= ? \\ h &= 6.63 \times 10^{-34} \text{ Js}^{-1} \\ e &= 1.6 \times 10^{-19} \text{ C} \\ V_0 &= 10^4 \text{ V} \end{aligned}$$

SOLUTION:

$$\begin{aligned} \lambda_{\min} &= \frac{hc}{eV_0} \\ \lambda_{\min} &= \frac{6.63 \times 10^{-34} \times 3 \times 10^8}{1.6 \times 10^{-19} \times 10^4} \\ \lambda &= 1.24 \times 10^{-10} \text{ m} \end{aligned}$$

17.10: In a Compton scattering process, the fractional change in wavelength of x-ray photon is 1% at an angle $\theta = 120^\circ$, find the wavelength of x-ray used in the experiment.

DATA:

$$\begin{aligned} \frac{\Delta \lambda}{\lambda_1} &= \frac{1}{100} \\ \lambda_1 &= 100 \\ \theta &= 120^\circ \\ \lambda_2 &= ? \end{aligned}$$

SOLUTION:

$$\begin{aligned} \frac{\Delta \lambda}{\lambda_1} &= \frac{1}{100} \\ \lambda_1 &= 100 \\ \Delta \lambda &= \frac{\lambda_1}{100} \end{aligned}$$

$$\begin{aligned} \Delta \lambda &= \frac{h}{m_0 C} (1 - \cos \theta) \\ \frac{\lambda_1}{100} &= \frac{6.63 \times 10^{-34}}{9.1 \times 10^{-31} \times (3 \times 10^8)} (1 - \cos 120^\circ) \\ \frac{\lambda_1}{100} &= 2.43 \times 10^{-12} \times 1.5 \\ \frac{\lambda_1}{100} &= 3.64 \times 10^{-12} \\ \lambda_1 &= 3.64 \times 10^{-12} \times 100 \\ \lambda_1 &= 3.64 \times 10^{-10} \text{ m} \end{aligned}$$

17.11: Find the wave length of 2.0g light ball moving with a velocity (a) 1.0mm per century (b) 1.0 mS^{-1} . (Given $h = 6.63 \times 10^{-34} \text{ Js}^{-1}$ and 1 year = $3.15 \times 10^7 \text{ S}$)

DATA:

$$\begin{aligned} \lambda &= ? \\ m_0 &= 2\text{g} = 2 \times 10^{-3} \text{ Kg} \\ \text{a) } V &= 1 \text{ mm per century} \\ \text{b) } V &= 1 \text{ m/s} \end{aligned}$$

SOLUTION:

$$\lambda = \frac{h}{m_0 V} \quad (1)$$

$$\begin{aligned} \text{a) } 1 \text{ mm} &= 1 \times 10^{-3} \text{ m} \\ 1 \text{ century} &= 3.15 \times 10^9 \text{ S} \\ V &= \frac{1 \times 10^{-3}}{3.15 \times 10^9} = 3.17 \times 10^{-13} \text{ m/s} \\ \text{eq (1)} \Rightarrow \lambda &= \frac{6.63 \times 10^{-34}}{2 \times 10^{-3} \times 3.17 \times 10^{-13}} \\ \lambda &= 1.05 \times 10^{-18} \text{ m} \\ \text{b) eq (1)} \Rightarrow \lambda &= \frac{6.63 \times 10^{-34}}{2 \times 10^{-3} \times 1} \\ \lambda &= 3.315 \times 10^{-31} \text{ m} \end{aligned}$$

17.12: An electron exist within a region of 10^{-10} m . Find its momentum uncertainty and the approximate kinetic energy.

DATA:

$$\begin{aligned} \Delta x &= 10^{-10} \text{ m} \\ \text{a) } \Delta p &= ? \\ \text{b) } \text{K.E} &= ? \end{aligned}$$

SOLUTION:

$$\begin{aligned} \text{a) } \Delta p \Delta x &= h \\ \Delta p &= \frac{h}{\Delta x} \\ \Delta p &= \frac{1.05 \times 10^{-34}}{10^{-10}} \end{aligned}$$

$$\Delta p = 1.05 \times 10^{-24} \text{ Kgms}$$

$$\begin{aligned} \Delta p &= \frac{h}{\Delta x} \\ \Delta p &= \frac{1.05 \times 10^{-34}}{10^{-10}} \\ \Delta p &= 1.05 \times 10^{-24} \text{ Kgms} \end{aligned}$$

$$\begin{aligned} \text{b) } \text{K.E} &= \frac{\Delta p^2}{2m_0} \\ \text{K.E} &= \frac{(1.05 \times 10^{-24})^2}{2 \times 9.1 \times 10^{-31} \times 1.6 \times 10^{-19}} \\ \text{K.E} &= 3.8 \text{ eV} \end{aligned}$$

17.13: Sodium surface is shined with light of wavelength 3×10^{-7} m. If the work function of Na = 2.46 eV, find the K.E of the Photoelectrons and also the cut off wave length.

DATA:

$$\begin{aligned} \lambda &= 3 \times 10^{-7} \text{ m} \\ \Phi &= 2.46 \text{ eV} \\ \text{K.E} &= ? \\ \lambda_c &= ? \end{aligned}$$

SOLUTION:

$$E = \frac{hc}{\lambda}$$

$$E = \frac{6.63 \times 10^{-34} \times 3 \times 10^8}{3 \times 10^{-7} \times 1.6 \times 10^{-19}}$$

$$\begin{aligned} E &= 4.14 \text{ eV} \\ \text{K.E} &= E - \Phi \\ \text{K.E} &= 4.14 - 2.46 \\ \text{K.E} &= 1.68 \text{ eV} \\ \Phi &= 2.46 \times 1.6 \times 10^{-19} \\ \Phi &= 3.936 \times 10^{-19} \text{ J} \\ \lambda_c &= \frac{hc}{\Phi} \\ \lambda_c &= \frac{6.63 \times 10^{-34} \times 3 \times 10^8}{3.936 \times 10^{-19}} \end{aligned}$$

$$\lambda_c = 5.05 \times 10^{-7} \text{ m}$$

17.14: X - Rays of wave length. λ_0 are scattered from a carbon black at an angle of 45° with respect to the incident beam. Find the shift in wave length.

DATA:

$$\begin{aligned} \text{given } \lambda_1 &= \lambda_0 \\ \text{then } \lambda_2 &= \lambda \\ \theta &= 45^\circ \\ \lambda - \lambda_0 &= ? \end{aligned}$$

SOLUTION:

$$\lambda - \lambda_0 = \frac{h}{m_0 c} (1 - \cos \theta)$$

$$\lambda - \lambda_0 = \frac{6.63 \times 10^{-34}}{9.1 \times 10^{-31} \times 3 \times 10^8} (1 - \cos 45^\circ)$$

$$\lambda - \lambda_0 = 7.11 \times 10^{-13} \text{ m}$$

17.15: If the electron beam in a TV Picture tube is, accelerated by 10,000V what will be the De'Broglie's wave length?

DATA:

$$\begin{aligned} V &= 10,000 \text{ V} \\ \lambda &= ? \end{aligned}$$

SOLUTION:

$$\lambda = \frac{h}{\sqrt{2m_0 eV}}$$

$$\lambda = \frac{6.63 \times 10^{-34}}{\sqrt{2 \times 9.110^{-31} \times 1.6 \times 10^{-19} \times 10,000}}$$

$$\lambda = 1.23 \times 10^{-11} \text{ m}$$

17.16: What minimum energy proton can be used to observe an object of size 2.5×10^{-10} m.

DATA:

$$\begin{aligned} \Delta E &= ? \\ \Delta x &= 2.5 \times 10^{-10} \text{ m} \end{aligned}$$

SOLUTION:

$$\begin{aligned} \Delta p \Delta x &= h \\ \Delta p &= \frac{h}{\Delta x} \\ \Delta p &= \frac{6.63 \times 10^{-34}}{2.5 \times 10^{-10}} \end{aligned}$$

$$\begin{aligned} \Delta p &= 2.652 \times 10^{-24} \text{ Kg m/s} \\ \Delta E &= mc^2 \\ \Delta E &= (mc)c \\ \Delta E &= \Delta pc \end{aligned}$$

$$\Delta p = \frac{2.652 \times 10^{-24} \times 3 \times 10^8}{1.6 \times 10^{-19}}$$

$$\Delta E = 4972.5 \text{ eV}$$

17.17: What will be the De'broglie's wavelength associated with a mass of 0.01 Kg moving with a Velocity 10 m/s.

DATA:

$$\begin{aligned}\lambda &= ? \\ m_0 &= 0.01 \text{ Kg} \\ V &= 10 \text{ m/s.}\end{aligned}$$

SOLUTION:

$$\begin{aligned}\lambda &= \frac{h}{m_0 V} \\ \lambda &= \frac{6.63 \times 10^{-34}}{0.01 \times 10} \\ \lambda &= 6.63 \times 10^{-33} \text{ m}\end{aligned}$$

17.18: Certain excited state of hydrogen atom have a life time 2.5×10^{-19} S. What will be the minimum uncertainty in energy?

DATA:

$$\begin{aligned}\Delta t &= 2.5 \times 10^{-19} \text{ S} \\ \Delta E &= ?\end{aligned}$$

SOLUTION:

$$\begin{aligned}\Delta E \Delta t &= h \\ \Delta E &= \frac{h}{\Delta t} \\ \Delta E &= \frac{1.05 \times 10^{-34}}{2.5 \times 10^{-19} \times 1.6 \times 10^{-19}} \\ \Delta E &= 2625 \text{ eV}\end{aligned}$$

17.19: X-Rays are scattered from a target material. The scattered radiation is viewed at an angle of 90° with respect to the incident beam. Find the Compton shift in wave length.

DATA:

$$\begin{aligned}\theta &= 90^\circ \\ \lambda_2 - \lambda_1 &= ?\end{aligned}$$

SOLUTION:

$$\begin{aligned}\lambda_2 - \lambda_1 &= \frac{h}{m_0 C} (1 - \cos \theta) \\ \lambda_2 - \lambda_1 &= \frac{6.63 \times 10^{-34}}{9.1 \times 10^{-31} \times 3 \times 10^8} (1 - \cos 90^\circ) \\ \lambda_2 - \lambda_1 &= 2.43 \times 10^{-12} \text{ m}\end{aligned}$$

17.20: Find the frequency of a photon when an electron of 20 KeV is brought to rest in a collision with a heavy nucleus.

DATA:

$$\begin{aligned}\nu &= ? \\ E &= 20 \text{ KeV}\end{aligned}$$

SOLUTION:

$$\begin{aligned}E &= 20 \times 10^3 \times 1.6 \times 10^{-19} \\ E &= 3.2 \times 10^{-15} \text{ J}\end{aligned}$$

$$E = h\nu$$

$$\nu = \frac{E}{h}$$

$$\nu = \frac{3.2 \times 10^{-15}}{6.63 \times 10^{-34}}$$

$$\nu = 4.83 \times 10^{18} \text{ Hz}$$

17.16 SOLVED NUMERICALS OF PAPERS:

YEAR 2013.

Q. Pair annihilation occurred due to a head-on-collision of an electron and a positron having the same kinetic energy producing pair of Photons each having 2.5MeV. What were their kinetic energies before collision?

(Given $m_0c^2 = 0.511$ MeV)

DATA:

$E = hv = 2.5$ MeV
 $E_0 = m_0c^2 = 0.511$ MeV
 $(K.E)^{e^-} = (K.E)^{e^+} = K.E$

SOLUTION:

As we know that for Annihilation of matter
 $2hv = 2m_0c^2 + (K.E)_{e^-} + (K.E)_{e^+}$
 Or $2hv = 2m_0c^2 + K.E + K.E$
 Or $2E = 2m_0c^2 + 2K.E$
 $2E = 2(m_0c^2 + K.E)$

Or $E = m_0c^2 + K.E$
 Or $K.E = E - m_0c^2$
 $= 2.5 - 0.511$
 $K.E = 1.989$ MeV
 $\therefore (K.E)_{e^-} = (K.E)_{e^+} = (K.E)$
 $(K.E)^{e^-} = (K.E)^{e^+} = 1.982$ MeV

YEAR 2011.

Q.2(xii) Find the shortest wavelength of photon emitted in the Balmer series and determine its energy in eV. ($R_H = 1.097 \times 10^{-7} m^{-1}$).

DATA:

Rest mass = m_0
 $K = 2E_0$
 (a) Velocity = $V = ?$
 (b) Momentum = $P = ?$

SOLUTION:

As given that
 $K = 2E_0$ -----(1)
 Using $[K = K.E]$
 $E = E_0 + 2E_0$
 Or $E = 3E_0$
 $mc^2 = 3m_0c^2$
 $m = 3m_0$ -----(2)

(a) For Velocity
 Equation (2) also written as

$$\frac{m_0}{\sqrt{1-\frac{v^2}{c^2}}} = 3m_0 \quad \left[m = \frac{m_0}{\sqrt{1-\frac{v^2}{c^2}}} \right]$$

$$\frac{1}{\sqrt{1-\frac{v^2}{c^2}}} = 3$$

$$\frac{1}{3} = \sqrt{1-\frac{v^2}{c^2}}$$

Squaring on both sides

$$\frac{1}{9} = 1 - \frac{v^2}{c^2}$$

Or $\frac{v^2}{c^2} = 1 - \frac{1}{9}$
 $\frac{v^2}{c^2} = \frac{9-1}{9}$
 $\frac{v^2}{c^2} = \frac{8}{9}$
 $v^2 = \frac{8}{9}c^2$

Or $\sqrt{v^2} = \sqrt{\frac{8}{9}c^2}$
 $v = \frac{\sqrt{8}}{3}c$ -----(A)

Or $v = \frac{2\sqrt{2}}{3}c$

Or $v = \frac{2 \times 1.44 \times 3 \times 10^8}{3}$
 $v = 2.828 \times 10^8$ m/sec

(b) For momentum

$P = mv$
 Put m and v from (2) and (A)

$P = 3m_0 \times \frac{\sqrt{8}}{3}c$

$P = \sqrt{8} m_0c$

Or $P = 2\sqrt{2} m_0c$

YEAR 2010.

Q.2(xiv) If the electron beam in a television picture tube is accelerated by 10000 volt, What will be the de-Broglie's hypothesis
 ($h = 6.63 \times 10^{-34} \text{ J sec}$) ($m = 9.1 \times 10^{-31} \text{ kg}$)

SOLUTION: Problem No. 17.15 of Book

ANSWER $\lambda = 1.23 \times 10^{-11} \text{ m}$

YEAR 2009.

Q.7 (d) A sodium surface is shined with the light of wavelength $3 \times 10^{-7} \text{ m}$. If work function of sodium metal is 2.46 eV. Find K.E of the photo electrons and also cut off wavelength.

SOLUTION:

SIMILAR TO QUESTION# 17.3

ANSWER (a) $K \cdot E = 1.68 \text{ eV}$

(b) $\lambda_c = \lambda_0 = 5.05 \times 10^{-7} \text{ m}$

YEAR 2008.

Q.7 (d) Calculate the relativistic speed at which mass of a particle becomes double its rest mass.

DATA:

$$m = 2m_0$$

$$\text{Relativistic speed} = v = ?$$

SOLUTION:

As we know that

$$m = \frac{m_0}{\sqrt{1 - \frac{v^2}{c^2}}}$$

$$\text{OR } 2m_0 = \frac{m_0}{\sqrt{1 - \frac{v^2}{c^2}}}$$

$$\sqrt{1 - \frac{v^2}{c^2}} = \frac{1}{2}$$

Squaring on both sides

$$1 - \frac{v^2}{c^2} = \frac{1}{4}$$

$$\text{OR } 1 - \frac{1}{4} = \frac{v^2}{c^2}$$

$$\frac{4-1}{4} = \frac{v^2}{c^2}$$

$$\text{OR } \frac{3}{4} = \frac{v^2}{c^2}$$

$$v^2 = \frac{3}{4} c^2$$

$$v^2 = \sqrt{\frac{3}{4}} c^2$$

$$\text{OR } v = \frac{\sqrt{3}}{2} c$$

$$\text{OR } v = \frac{1.732}{2} \times 3 \times 10^8 \text{ m/sec}$$

$$\boxed{v = 2.598 \times 10^8 \text{ m/sec}}$$

YEAR 2007.

Q.8 (d) In Compton scattering process the fractional change in wavelength of x-rays photon is 1% at an angle 120°, find the wavelength of x-rays used in this experiment.

DATA:

$$\text{Fractional change} = \frac{\Delta\lambda}{\lambda_1} = 1\% = \frac{1}{100} = 10^{-2}$$

$$\text{Angle} = \theta = 120^\circ$$

$$\text{Wavelength of x-rays} = \lambda_1 = ?$$

SOLUTION:

$$\text{As } \Delta\lambda = \lambda_c(1 - \cos\theta) \text{ ————— (1)}$$

Where

$$\lambda_c = \frac{h}{m_0c}$$

$$\lambda_c = \frac{6.63 \times 10^{-34}}{9.11 \times 10^{-31} \text{ kg} \times 3 \times 10^8} = \frac{6.63 \times 10^{-34}}{27.33 \times 10^{-23}}$$

$$\lambda_c = 2.426 \times 10^{-12} \text{ m}$$

$$\therefore (1) \Rightarrow \Delta\lambda = 2.426 \times 10^{-12} (1 - \cos 120^\circ)$$

$$\Delta\lambda = 2.426 \times 10^{-12} [1 - (-0.5)]$$

$$\Delta\lambda = 2.426 \times 10^{-12} \times 1.5$$

$$\frac{\Delta\lambda}{\lambda_1} = \frac{3.6389 \times 10^{-12}}{\lambda_1}$$

$$10^{-2} = \frac{3.639 \times 10^{-12}}{\lambda_1}$$

$$\lambda_1 = \frac{3.639 \times 10^{-12}}{10^{-2}}$$

$$\boxed{\lambda_1 = 3.639 \times 10^{-10} \text{ m}}$$

$$[\because 10^{-10} \text{ m} = 1 \text{ \AA}]$$

$$\text{OR } \boxed{\lambda_1 = 3.64 \text{ \AA}}$$

YEAR 2006.

Q.7(d) An electron exists within a region 10^{-10} m find its momentum uncertainty and approximate kinetic energy.

SOLUTION:

SIMILAR TO QUESTION# 17.12

ANSWER (a) $\Delta P = 1.05 \times 10^{-24} \text{ N-sec}$

(b) $K.E = 3.8 \text{ eV}$

YEAR 2005.

Q.8 (d) Estimate the relativistic mass and the wavelength associated with an electron moving at 0.9 C ($h = 6.63 \times 10^{-34}$ J - sec).

DATA:

Rest mass of electron = $m_0 = 9.11 \times 10^{-31}$ kg

Velocity of electron = $v = 0.9$ C

Plank's constant = $h = 6.63 \times 10^{-34}$ J-sec

- (a) Relativistic mass of electron = $m = ?$
(b) Wavelength = $\lambda = ?$

SOLUTION:

As we know that

$$m = \frac{m_0}{\sqrt{1 - \left(\frac{v}{c}\right)^2}}$$

OR
$$m = \frac{9.11 \times 10^{-31}}{\sqrt{1 - \left(\frac{0.9c}{c}\right)^2}}$$

$$m = \frac{9.11 \times 10^{-31}}{\sqrt{1 - 0.81}}$$

$$m = \frac{9.11 \times 10^{-31}}{0.43589}$$

$$m = 2.0899 \times 10^{-30} \text{ kg}$$

Using De-Broglie Hypothesis

$$\lambda = \frac{h}{mv}$$

$$\lambda = \frac{6.63 \times 10^{-34}}{2.0899 \times 10^{-30} \times 0.9 \times 3 \times 10^8}$$

$$\lambda = \frac{6.63 \times 10^{-34}}{5.6427 \times 10^{-22}}$$

$$\lambda = 1.175 \times 10^{-12} \text{ m}$$

OR
$$\lambda = 1.175 \times 10^{-2} \times 10^{-10} \text{ m}$$

$$\lambda = 1.175 \times 10^{-2} \times \text{\AA}$$

YEAR 2004.

Q.8 (d) Compare the energy of photon of wavelength 2×10^{-6} m with the energy of x - ray photon of wavelength 2×10^{-10} m ($h = 6.63 \times 10^{-34}$ J-sec)

DATA:

Wavelength of photon = $\lambda_p = 2 \times 10^{-6}$ m

Wavelength of x - ray = $\lambda_x = 2 \times 10^{-10}$ m

Plank's constant = $h = 6.63 \times 10^{-34}$ J-sec

Compare the energy = $E_p : E_x = \frac{E_p}{E_x} = ?$

SOLUTION:

As we know that

$$E = h\nu$$

OR
$$E = \frac{hc}{\lambda} \text{ --- (1)}$$

For Photon

$$E_p = \frac{hc}{\lambda_p} \text{ --- (2)}$$

For x - ray

$$E_x = \frac{hc}{\lambda_x} \text{ --- (3)}$$

Now
$$E_p : E_x = \frac{E_p}{E_x}$$

$$E_p : E_x = \frac{hc}{\lambda_p} \div \frac{hc}{\lambda_x}$$

$$E_p : E_x = \frac{hc}{\lambda_p} \times \frac{\lambda_x}{hc}$$

$$E_p : E_x = \frac{\lambda_x}{\lambda_p}$$

$$E_p : E_x = \frac{2 \times 10^{-10}}{2 \times 10^{-6}}$$

$$E_p : E_x = 10^{-4}$$

OR
$$\frac{E_p}{E_x} = 10^{-4}$$

YEAR 2004.

Q.8 (d) 2nd Method

As $E = h\nu$

OR $E = \frac{hc}{\lambda}$ ————— (1) $\left[\begin{array}{l} \nu = \frac{c}{\lambda} \\ c = 3 \times 10^8 \text{ m/sec} \end{array} \right]$

FOR PHOTON

$$E_p = \frac{hc}{\lambda_p}$$

$$E_p = \frac{6.63 \times 10^{-34} \times 3 \times 10^8}{2 \times 10^{-6}}$$

$$E_p = 9.945 \times 10^{-20} \text{ J} \text{ ————— (2)}$$

FOR X-RAY

(1) $\Rightarrow E_x = \frac{hc}{\lambda_x}$

$$E_x = \frac{1.989 \times 10^{-25}}{2 \times 10^{-10}}$$

$$E_x = 9.945 \times 10^{-16} \text{ J} \text{ ————— (3)}$$

Now $E_p : E_x = \frac{E_p}{E_x}$

$$E_p : E_x = \frac{9.945 \times 10^{-20}}{9.945 \times 10^{-16}}$$

$$E_p : E_x = 10^{-4}$$

OR

$$\frac{E_p}{E_x} = 10^{-4}$$

YEAR 2002.P.M.

Sodium surface is shined with the light of wave length $3 \times 10^{-7} \text{ m}$. If the work function of Na = 2.46 eV, find the Kinetic energy of the photo electrons. [C = $3 \times 10^8 \text{ m/s}$; h = $6.63 \times 10^{-34} \text{ J-S}$].

DATA:

$$\lambda = 3 \times 10^{-7} \text{ m}$$

$$\phi = 2.46 \text{ eV}$$

$$K = ?$$

SOLUTION:

$$E = \frac{hc}{\lambda}$$

$$E = \frac{6.63 \times 10^{-34} \times 3 \times 10^8}{3 \times 10^{-7} \times 1.6 \times 10^{-19}}$$

$$E = 4.14 \text{ eV}$$

$$K = E - \phi$$

$$K = 4.14 - 2.46$$

$$K = 1.68 \text{ eV}$$

YEAR 2002.P.M.

Find the speed at which the mass of a particle will be doubled. [$c = 3 \times 10^8$ m/s]

SOLUTION:

$$m = 2m_0$$

$$\frac{m_0}{\sqrt{1 - \frac{V^2}{C^2}}} = 2m_0$$

$$\frac{1}{2} = \sqrt{1 - \frac{V^2}{C^2}}$$

$$\frac{1}{4} = 1 - \frac{V^2}{C^2}$$

$$v = \frac{\sqrt{3}}{2}c$$

$$v = 0.8666 \times 3 \times 10^8$$

$$\boxed{v = 2.6 \times 10^8 \text{ m/s}}$$

YEAR 2002.P.E.

Given $m_0c^2 = 0.511$ MeV. Find the total energy E and the Kinetic energy K of an electron moving with a speed $v = 0.85c$.

(Rest mass of electron = 9.1×10^{-31} Kg and velocity of light is 3×10^8 m/s).

DATA:

$$m_0c^2 = E_0 = 0.511 \text{ MeV}$$

$$v = 0.85c$$

$$m_0 = 9.1 \times 10^{-31} \text{ Kg}$$

$$c = 3 \times 10^8 \text{ m/s}$$

$$E = ?, \quad K = ?$$

SOLUTION:

$$m = \frac{m_0}{\sqrt{1 - \frac{V^2}{C^2}}}$$

$$mc^2 = \frac{m_0c^2}{\sqrt{1 - \frac{V^2}{C^2}}}$$

$$E = \frac{E_0}{\sqrt{1 - \frac{V^2}{C^2}}}$$

$$E = \frac{0.511 \text{ MeV}}{\sqrt{1 - \left(\frac{0.85c}{c}\right)^2}}$$

$$\boxed{E = 0.97 \text{ MeV}}$$

$$K = E - E_0$$

$$K = 0.97 - 0.511$$

$$\boxed{K = 0.459 \text{ MeV}}$$

17.17 NUMERICALS FOR SELF PRACTICE:

YEAR 2012.

Q.2(iv) Given $m_0c^2 = 0.511\text{MeV}$. Find the total energy "E" and the kinetic energy "K" of an electron moving with a speed $0.85c$

$$(m_0 = 9.1 \times 10^{-31} \text{Kg}, c = 3 \times 10^8 \text{ m/Sec}).$$

Ans. **Q#17.3 of Book**

Q.2(xv) A sodium surface is exposed to a light of wavelength $3 \times 10^{-7}\text{m}$. If the work function of sodium metal is 2.46eV find the K.E of the photoelectrons and cut-off wavelength.

Ans. **Q#17 of Book**

YEAR 1999.

The range of visible light is 4000\AA to 7000\AA , will photo-electrons be emitted by a copper surface of work function 4.4eV , when illuminated by visible light?

Ans. **$E < \phi$**

YEAR 1994.

The work function of a metal is 2eV . The light of wavelength 3000\AA is made to fall on it. Find the Kinetic energy of the fastest emitted photoelectrons.

Ans. **$\text{K.E} = 2.14 \text{ eV}$**

YEAR 1992.

The work function of a photo emissive surface is 4.0 eV . What will be the velocity of the fastest photo electrons emitted from it by an incident light of frequency $3 \times 10^{15} \text{ Hz}$. ($h=6.63 \times 10^{-34} \text{ J. sec}$).

Ans. **$v = 1.72 \times 10^6 \text{ m/s}$**

YEAR 1989.

The work function of a certain metal is 3.0 eV , when this metal is illuminated by the infra-red light of $1.2 \times 10^{15} \text{ Hertz}$, find the maximum K.E. of the emitted photo electrons. ($h=6.63 \times 10^{-34} \text{ J. sec}$).

Ans. **$\text{K.E} = 1.97\text{eV}$**

YEAR 1987.

Find the cut-off wave length for a given metal whose work function is 4.4 eV . ($h=6.63 \times 10^{-34} \text{ J. sec}$).

Ans. **$\lambda_c = 282.2 \text{ nm}$**